



Effect of NiCr Clad BaF₂·CaF₂ Addition on Wear Performance of Plasma Sprayed Chromium Carbide-Nichrome Coating

Lingzhong Du, Chuanbing Huang, Weigang Zhang, Jingmin Zhang, and Wei Liu

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NiCr clad BaF₂·CaF₂ fluoride eutectic powders were added into chromium carbide-nichrome feedstock to improve the tribological properties of NiCr-Cr₃C₂ coating, and the structures, mechanical, and ball-on-disk sliding wear performance of the coating were characterized. The results show that NiCr cladding can effectively decrease the density and thermophysical difference between the feedstock components, while alleviate the decarburization and oxidization of the constituent phases, and form the coating with a uniform and dense microstructure. However, the addition of BaF₂·CaF₂ has a negative effect on mechanical properties of the coating. When the temperature reaches 500 °C, the BaF₂·CaF₂ eutectic is softened by the heat and smeared by the counterpart, thus the low shear stress lubricating film forms between the contact surface, that improves the tribological properties dramatically. At this temperature, the dominant wear mechanisms also change from splats spallation and abrasive wear at room temperature to plastic deformation and plowing by the counterpart. Within the temperature range from 600 to 800 °C, the friction coefficient, the wear rates of NiCr/Cr₃C₂-10% BaF₂·CaF₂ coating and its coupled Si₃N₄ ball are 20%, 40%, and 75% lower than those of the NiCr/Cr₃C₂ coating, respectively. The NiCr/Cr₃C₂-BaF₂·CaF₂ coating shows superior wear performance to the NiCr/Cr₃C₂ coating without lubricant additive.

Keywords metal-matrix composite, solid lubricant coating, sliding wear, thermal spray coating, wear testing

1. Introduction

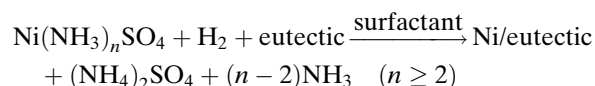
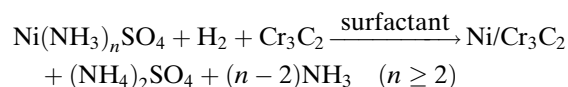
Thermal sprayed chromium carbide-nichrome coatings are generally produced for wear applications at the elevated temperature involving sliding, fretting, abrasion, and erosion in energy conversion and utilization systems like steam turbines, gas turbines, etc. (Ref 1-9). However, the high friction coefficient always leads the severe wear of the couple parts. BaF₂·CaF₂ fluoride eutectic with a much lower melting point (1050 °C) than BaF₂ (1338 °C) and CaF₂ (1410 °C) and with lower shear stress shows good lubrication under the temperature ranging between 400 and 900 °C and has been blended with NiCr and Cr₃C₂ powder to improve the wear properties of thermal sprayed NiCr-Cr₃C₂ coating (Ref 10, 11). However, the feedstock material prepared by a simple mechanically mixed NiCr, Cr₃C₂, and BaF₂·CaF₂ components has many serious disadvantages. Different density and thermophysical

properties of each constituent phase usually lead to separate trajectories and poor microstructure in the final coating. Also, the oxidation and decarburization of Cr₃C₂ powders during thermal spray result in poor wear resistance of the coating (Ref 12). In this study, a new feedstock preparation method by blending NiCr-coated BaF₂·CaF₂ fluoride eutectic and NiCr-coated Cr₃C₂ powders is used to overcome the shortcomings mentioned above, and the effect of the BaF₂·CaF₂ eutectic addition on the mechanical and wear performance of the plasma sprayed chromium carbide coatings is also investigated.

2. Materials and Experimental Procedures

2.1 Preparation of Feedstock and Coatings

Ni-coated Cr₃C₂ (Ni/Cr₃C₂) and Ni-coated BaF₂·CaF₂ (62:38 wt.%) eutectic (Ni/BaF₂·CaF₂) powders were, respectively, manufactured by the process of hydrothermal hydrogen reduction, as follows:



In the above reaction, the metallic nickel was reduced onto core particles from ammoniacal solution by hydrogen

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under pressure at a temperature of 150 °C in autoclaves. Then, the Ni/Cr₃C₂ and Ni/BaF₂·CaF₂ were, respectively, chromized to form NiCr-coated composite powders (NiCr/Cr₃C₂ and NiCr/BaF₂·CaF₂). The feedstocks with the required lubricant content were prepared by blending the NiCr/Cr₃C₂ (Ni:Cr:Cr₃C₂ = 20:5:75, wt.%) and NiCr/BaF₂·CaF₂ (Ni:Cr:BaF₂·CaF₂ = 40:10:50, wt.%) powders at the calculated ratio, as summarized in Table 1. The NiCr clad Cr₃C₂ powders without added lubricants functioned as the control or baseline coating to which the modified coatings were compared.

Before coating preparation, inconel X-750 superalloy substrates with a size of 30 × 40 mm² were grit blasted with coarse Al₂O₃ particles (about 600 μm) to roughen the surface, and then deposited NiAl for bond coating to improve the adhesion of the coating. The bond and top coatings were deposited by a commercial APS2000 K plasma spray system (Aeronautical Manufacturing Technology Research Institute, China) at optimal spray parameters, as summarized in Table 2. The thickness of NiAl bond coat and top composite layer was around 0.1 and 0.5 mm, respectively.

2.2 Coating Characterization

Vickers microhardness indentations (Shanghai Taiming Optical, China) with a load of 1.96 N and a dwell time of 15 s were made onto the polished cross section of the as-sprayed coatings to avoid the influence of the substrate. The microhardness of the coatings represented average of 10 indentations. Bond strength of the as-sprayed coatings was evaluated using a pull-off adhesion bond method as ASTM standard C-633. The crystal phases were characterized by an X'Pert Pro X-ray diffraction meter (Panalytical, Netherlands) with a Cu K α radiation.

2.3 Tribological Test

Friction and wear tests were conducted by HT-1000 “ball-on-disk” high temperature wear test equipment

Table 1 Blended ratio of NiCr/Cr₃C₂ and NiCr/BaF₂·CaF₂ powders in the feedstock

25NiCr/75Cr ₃ C ₂ ratio, wt.%	50NiCr/50BaF ₂ ·CaF ₂ ratio, wt.%	BaF ₂ ·CaF ₂ , wt.%
100	0	0
90	10	5
80	20	10
70	30	15
60	40	20

Table 2 Thermal spray parameters

System	Voltage, V	Current, A	Ar pressure, MPa	Ar flow rate, L/min
APS2000K	60	600	0.6	35
H ₂ pressure, MPa	H ₂ flow rate, L/min	Carrier gas pressure, N ₂ , MPa	Powder feed rate, g/min	Spray distance, mm
0.45	2	0.3	45	90

(Lanzhou Institute of Chemical Physics, China) under dry sliding condition in the air. The lower disk specimen of inconel X-750 superalloy was rotating, with a dimension of Ø25 × 5 mm². The thickness of composite coating deposited onto the superalloy disk was about 0.5 mm after grinding and polishing. The mating upper sintered Si₃N₄ ceramic ball (Shanghai Research Institute of Materials, China) was stationary, with a diameter of Ø=6 mm. Before the test, the disk and Si₃N₄ ceramic ball were cleaned by acetone and ultrasonic. The wear couples were surrounded by an electric resistance furnace and heated to the pre-set test temperature at a rate of 10 K/min. The temperature deviation was controlled to be within ±2 K. All tests were commenced at a load of 10 N with a sliding diameter of 10 mm and a sliding velocity of 0.188 m/s for duration of 20 min. The friction coefficient was recorded during each test and the mean values were calculated.

The cross section area of the worn track was measured using Talysurf 5P-120 surface profilometer (Rank Taylor Hobson, UK), and the area was multiplied by the track circumference to determine the disk wear volume. The wear volume of the counterpart was calculated by measuring the circular wear scar on the ball surface. The specific wear rates were obtained by dividing the wear volume by the load and the total sliding distance. To insure the reproducibility of the measurements, three repeated tests were executed under the same test conditions. The microstructures were investigated by scanning electron microscope (SEM), as well as chemical qualitative and semiquantitative analysis by energy-dispersive spectrometer (EDS).

3. Results and Discussion

3.1 Microstructure

The SEM image in Fig. 1(a) shows that the composite powders consist of blended NiCr/Cr₃C₂ (80 wt.%) and NiCr/BaF₂·CaF₂ (20 wt.%) with angular shape and size distribution between 45 and 75 μm. The cross section of the NiCr/BaF₂·CaF₂ (Fig. 1b) in backscattered electron image reveals that fluoride eutectic has a laminar structure, in which rounded CaF₂ grains (dark region in A) are uniformly distributed in the BaF₂ matrix (bright region in A) as observed by the other researcher (Ref 13). From the images, it is obviously that all of Cr₃C₂ (Fig. 1c) and BaF₂·CaF₂ particles have been completely and densely coated within the NiCr metallic binder. It is expected to

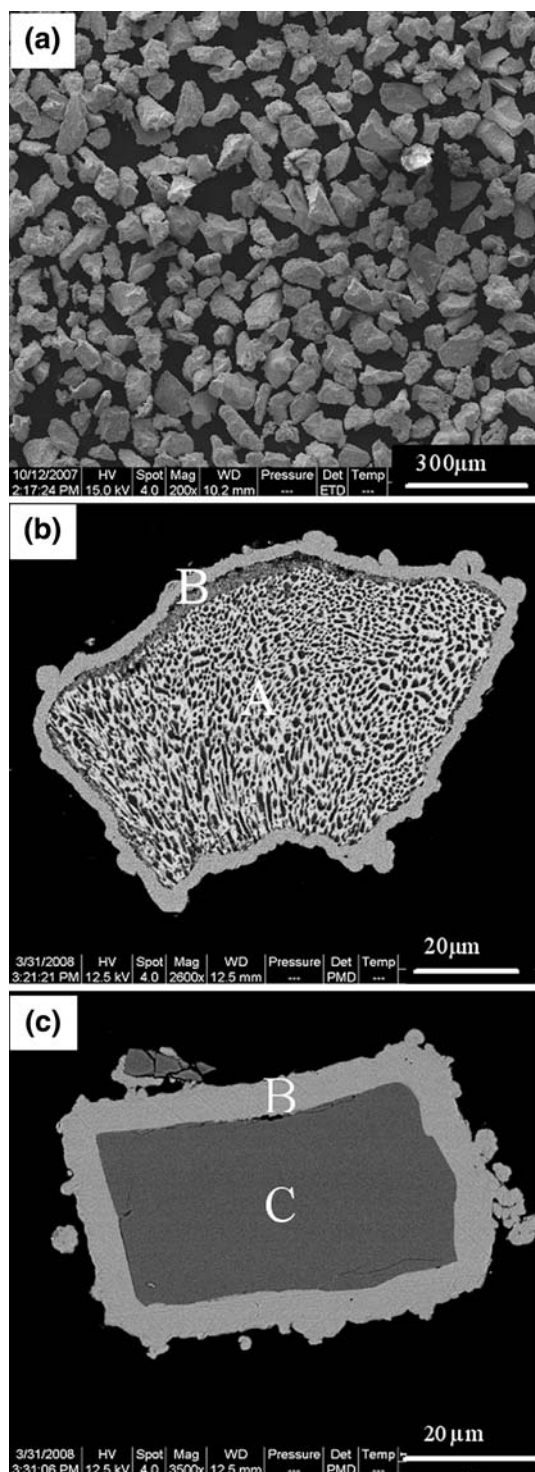


Fig. 1 SEM micrographs of the feedstock (80 wt.% NiCr/Cr₃C₂ blended with 20 wt.% NiCr/BaF₂·CaF₂) (a) blended powder surface, (b) NiCr/BaF₂·CaF₂ cross section, (c) NiCr/Cr₃C₂ cross section. Regions A, B, and C correspond to BaF₂·CaF₂ eutectic, NiCr alloy, and Cr₃C₂, respectively

prevent decarburization and oxidation of the cores, as well as to produce dense coating by melting of the surrounding binder during the thermal spraying process.

Table 3 Properties of the feedstock

	Flowability, s/50 g	Apparent density, g/cm ³
50NiCr/50BaF ₂ ·CaF ₂	59	1.85
25NiCr/75Cr ₃ C ₂	62	1.92

As summarized in Table 3, the NiCr/BaF₂·CaF₂ powders with 50 wt.% NiCr and NiCr/Cr₃C₂ powders with 25 wt.% NiCr have almost the same apparent density and the flowability, which expects to eliminate the negative effect of the separation of the blended feedstock in particle trajectories in the plasma jet.

Cross-sectional SEM images of the as-sprayed coatings with 0%, 10%, and 20% BaF₂·CaF₂ in the powders are shown in Fig. 2. Relatively dense coatings are well bonded with the substrates and lamellae microstructures elongated in the direction parallel to surface are observed, typically seen in plasma spray coating (Ref 14, 15). The porosity of the coatings estimated by image analysis increases from 3.63%, 4.73% to 5.42% with increase of the BaF₂·CaF₂ ratio. An EDS examination of the microstructure of the coating with 10% BaF₂·CaF₂ content in the powders reveals the darkest spot, marked by A in Fig. 4(d), corresponds to the pores. Also, the gray angular field of B, the bright gray area of C, and the broken particles of D correspond to Cr₃C₂, NiCr and BaF₂·CaF₂ fluoride, respectively.

X-ray diffraction patterns of the starting powder and the as-sprayed coating with 10% BaF₂·CaF₂ content in the powders are shown in Fig. 3. Before and after the spraying, all peaks of NiCr, Cr₃C₂, BaF₂, and CaF₂ are found in the XRD patterns, and no obvious peaks from Cr₇C₃ and Cr₂₃C₆ are detected in the coating. The decarburization of Cr₃C₂ to Cr₇C₃ or Cr₂₃C₆ is associated with oxidation of Cr₃C₂ by heating of sprayed particles (Ref 16). It is expected that densely coated NiCr alloy can effectively prevent oxidation, thus the decarburization is lower. It is common knowledge now that low decarburization will markedly improve the wear resistance of the Cr₃C₂ coating.

3.2 Mechanical Properties

The effect of the BaF₂·CaF₂ eutectic addition on the mechanical properties of the coatings are shown in Fig. 4. The Vickers microhardness in Fig. 4(a) and bond strength in Fig. 4(b) decrease slightly when the content of BaF₂·CaF₂ eutectic in the powders is <10%. When the BaF₂·CaF₂ eutectic exceeds 10%, they both drop sharply. It is evident that increase of the BaF₂·CaF₂ eutectic in the powders has a negative influence on the microhardness and bond strength of the sprayed coatings, which may result from the soft nature of the BaF₂·CaF₂ eutectic (eutectic HV_{0.2} = 450, Cr₃C₂ HV_{0.2} = 1400) and poor bond strength of BaF₂·CaF₂ eutectic when it melt in the plasma flame and broke during colliding on the substrate, as observed in Fig. 4(d).

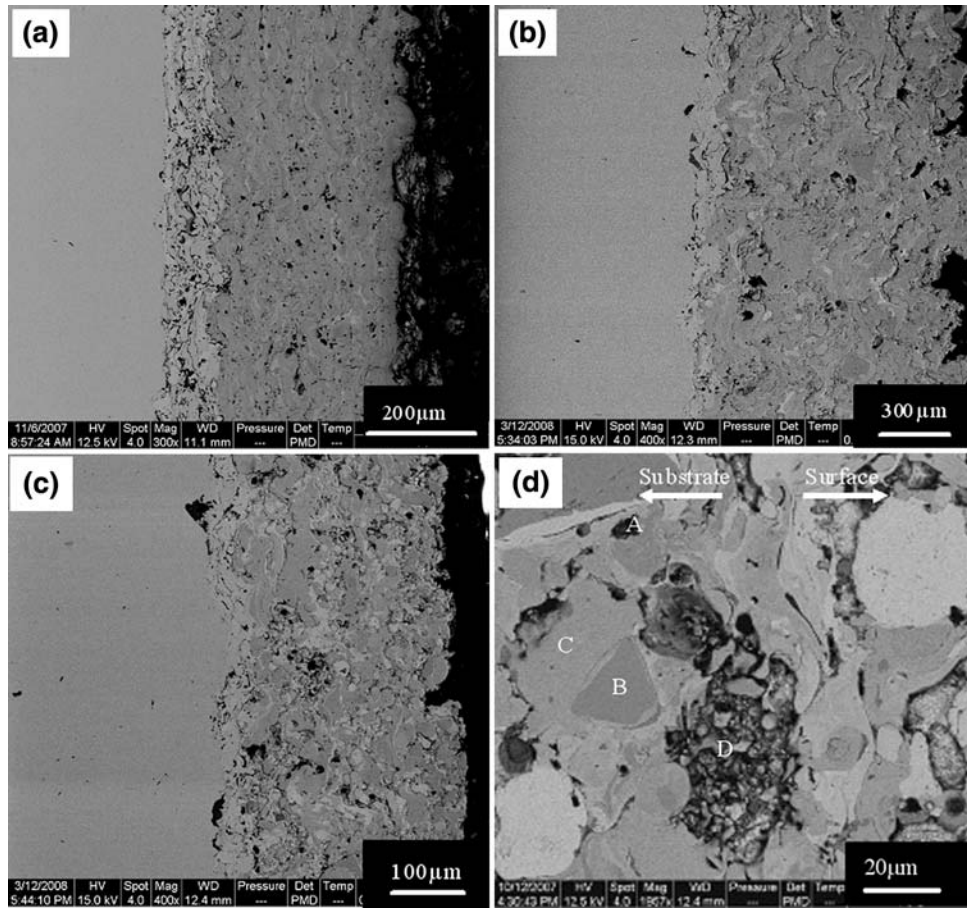


Fig. 2 SEM micrographs of the cross section of the as-sprayed NiCr/Cr₃C₂-BaF₂-CaF₂ coatings, (a) with no BaF₂-CaF₂, (b) with 10% BaF₂-CaF₂, (c) with 20% BaF₂-CaF₂ in the powders, and (d) higher magnification micrograph of (b)

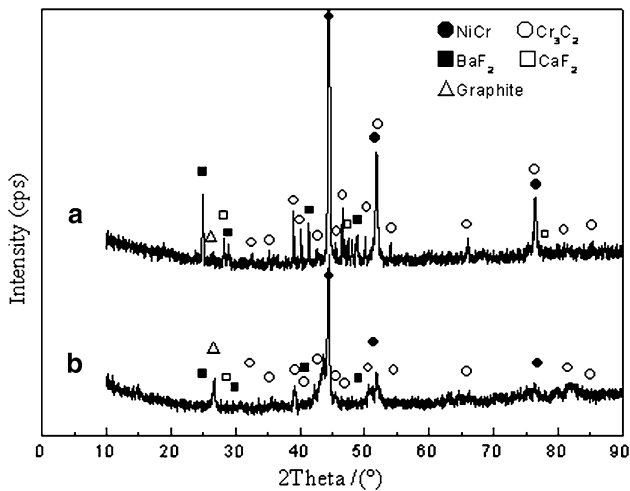


Fig. 3 XRD patterns of the composite NiCr/Cr₃C₂-BaF₂-CaF₂ powder with 10% BaF₂-CaF₂ addition and the as-sprayed coating

As is well known, the microhardness and bond strength are two vital important factors for the wear performance of the coating and too low microhardness and bond

strength will lead to the poor wear resistance. Therefore, the coating of 10% BaF₂-CaF₂ eutectic in the powders (defined as NiCr/Cr₃C₂-10%BaF₂-CaF₂) is selected to test its tribological properties.

3.3 Tribological Properties

Typical friction coefficients between wear couples with sliding distance at 600 °C and steady friction coefficients from room temperature to 800 °C of the NiCr/Cr₃C₂ coating and the NiCr/Cr₃C₂-10%BaF₂-CaF₂ coating are shown in Fig. 5. The friction coefficients of both coatings (Fig. 5a) increase gradually with sliding distance due to the running-in at the start of the test, and then decrease to relatively steady value. The friction coefficient and its fluctuations of NiCr/Cr₃C₂-10%BaF₂-CaF₂ coating are lower than those of NiCr/Cr₃C₂ coating. As the temperature increases, the friction coefficients of both NiCr/Cr₃C₂ and NiCr/Cr₃C₂-10%BaF₂-CaF₂ coatings increase below 300 °C, and then decrease. From room temperature to 500 °C, the friction coefficient of the NiCr/Cr₃C₂-10%BaF₂-CaF₂ coating is only about 5% lower than that of NiCr/Cr₃C₂ coating. When the temperature exceeds 500 °C, the friction coefficient of the NiCr/Cr₃C₂-10%BaF₂-CaF₂

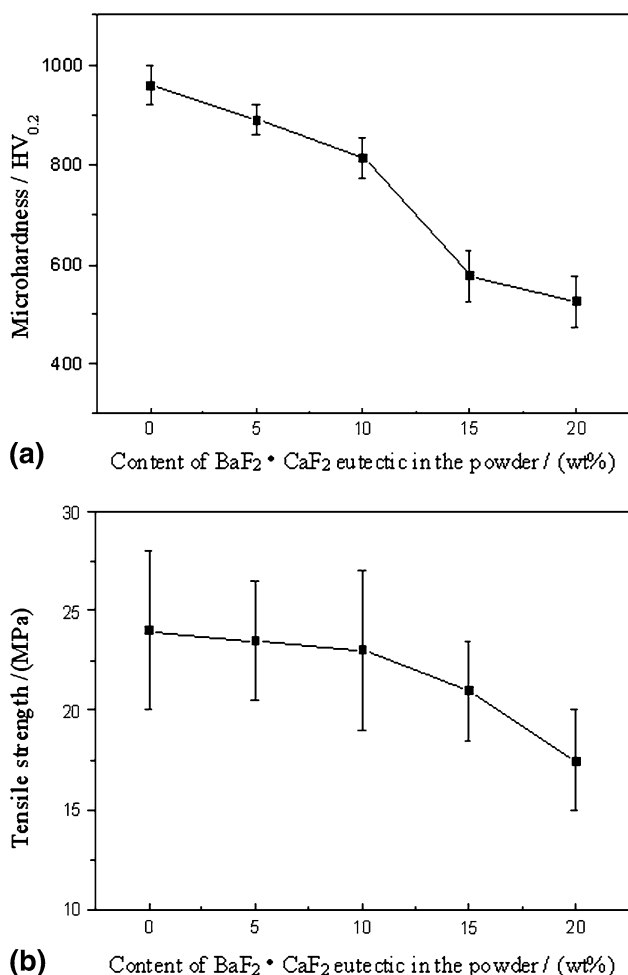


Fig. 4 Microhardness (a) and bond strength (b) of the NiCr/Cr₃C₂-BaF₂-CaF₂ coatings with the content of BaF₂-CaF₂ in the powders

coating has a notable reduction. It is about 0.4-0.5, and is 20% lower than that of the NiCr/Cr₃C₂ coating in the temperature region between 500 and 800 °C.

The specific wear rates of the NiCr/Cr₃C₂ coating, the NiCr/Cr₃C₂-10%BaF₂-CaF₂ coating and their counterpart Si₃N₄ balls from room temperature to 800 °C are shown in Fig. 6. Below 500 °C, the wear rates of both NiCr/Cr₃C₂ and NiCr/Cr₃C₂-10%BaF₂-CaF₂ coatings increase with increasing temperature, where the wear rate of the NiCr/Cr₃C₂ coating is lower than that of the NiCr/Cr₃C₂-10%BaF₂-CaF₂ coating. When the temperature exceeds 500 °C, the wear rate of the NiCr/Cr₃C₂ coating changes little, whereas the wear rate of the NiCr/Cr₃C₂-10%BaF₂-CaF₂ coating decreases from 10×10^{-5} to 5×10^{-5} mm³/N m. In the temperature region between 500 and 800 °C, the wear rate of the NiCr/Cr₃C₂-10%BaF₂-CaF₂ coating exhibits 40% lower than that of the NiCr/Cr₃C₂ coating. From Fig. 6(b), it can be seen that the wear rate of the Si₃N₄ ball for the NiCr/Cr₃C₂ coating increases at first with increasing temperature, and

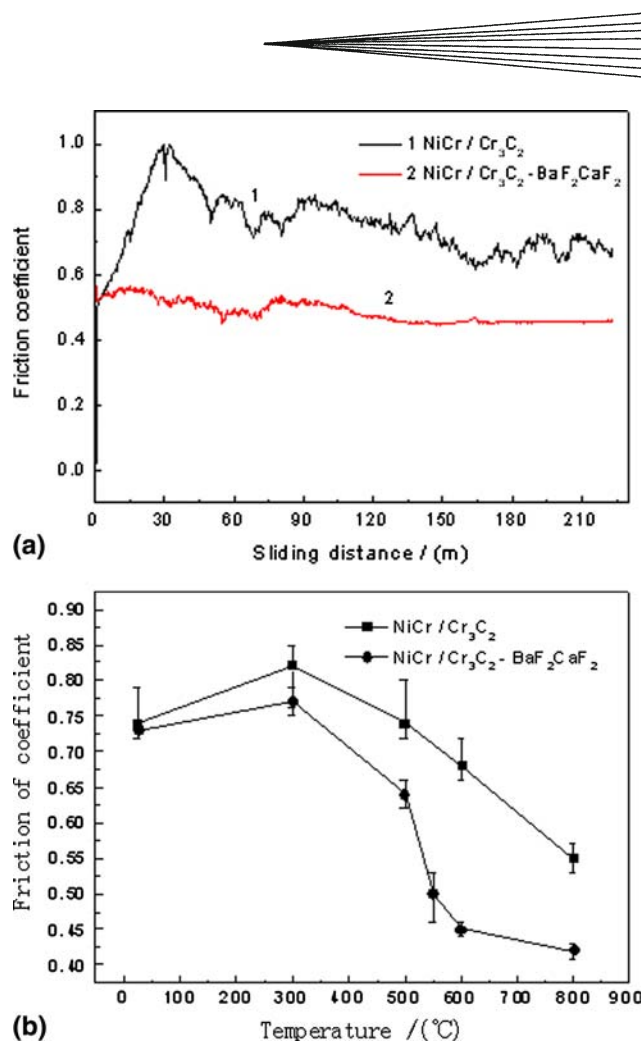
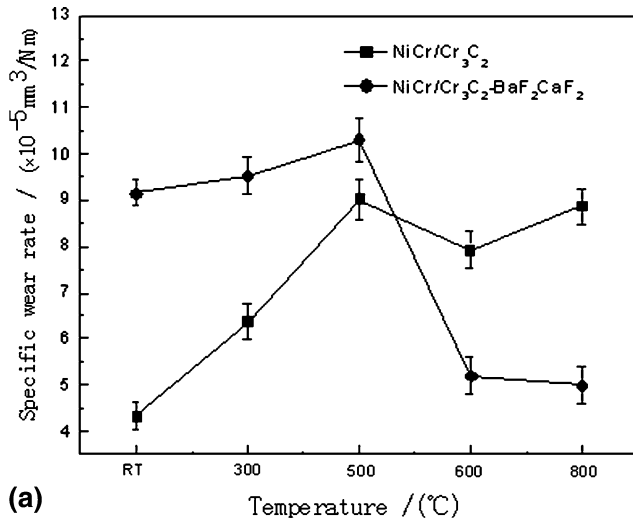


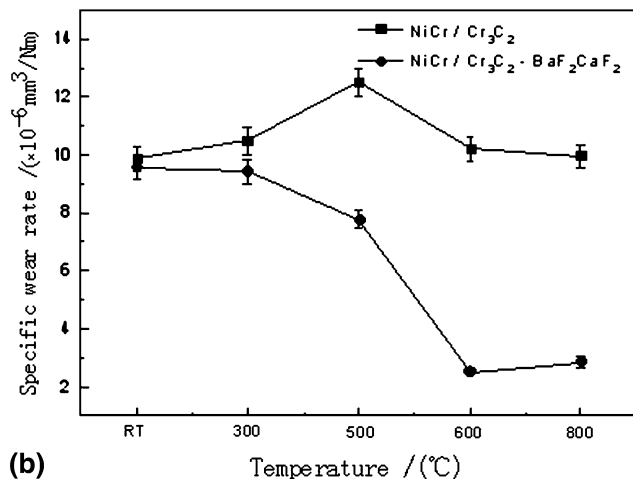
Fig. 5 Typical friction coefficients between wear couples with sliding distance at 600 °C (a) and steady friction coefficients from room temperature to 800 °C (b) of the NiCr/Cr₃C₂ coating and the NiCr/Cr₃C₂-10%BaF₂-CaF₂ coating

then decreases. On the other hand, the wear rate of the NiCr/Cr₃C₂-10%BaF₂-CaF₂ coating decreases monotonously. At the temperature from 600 to 800 °C, the wear rate of the Si₃N₄ ball for the NiCr/Cr₃C₂-10%BaF₂-CaF₂ coating is 75% lower than that for the NiCr/Cr₃C₂ coating.

Within the study, the tribological properties of NiCr/Cr₃C₂ coating improve by addition of the BaF₂-CaF₂ under the temperature of higher than 500 °C. To investigate the wear mechanism, SEM images of worn surface for the NiCr/Cr₃C₂-10% BaF₂-CaF₂ coating at room temperature and 500 °C are shown in Fig. 7(a) and (b). The worn surface of the coating at room temperature shows a rough surface with tiny wear scars grooved in the wear track. Under the continuous friction force, the splats will fracture and spalled at the weak regions, such as microcracks and pores. The spalled debris then acts as hard particles to plow the coating. Therefore at the room temperature, delamination and abrasive wear are considered to be two dominant wear mechanisms. With the temperature reaching up to 500 °C, the worn surface of the coating



(a)



(b)

Fig. 6 Specific wear rates with temperature of the NiCr/Cr₃C₂ coating and the NiCr/Cr₃C₂-10%BaF₂CaF₂ coating (a) and their counterpart Si₃N₄ balls (b)

shown in Fig. 7(b) becomes smooth and is covered by a glaze layer. EDS analysis (Fig. 7c) confirms that the layer mainly consists of O, Cr, C, Ba, Ca, and F with a little Si and Ni. It is obvious that a solid lubricious film selectively builds up on the sliding contact surface when the BaF₂·CaF₂ eutectic is softened at high temperature and smeared by the Si₃N₄ ball. This low shear stress lubricious film could effectively prevent fracture and spallation. At this temperature, the wear loss would be dominated by plastic deformation under pressure contact and then plowing by the counterpart. Based on the results, it is clear that the NiCr cladding can effectively reduce the difference of the density and thermophysical properties between the feedstock component and protect the cores from oxidation and decarburization. NiCr/Cr₃C₂ coating of 10% BaF₂·CaF₂ eutectic in the powders shows low friction coefficient and high wear resistance behavior under the temperature above 500 °C. It could be a promising way to improve the high temperature wear resistance by adding NiCr clad BaF₂·CaF₂ eutectic in NiCr/Cr₃C₂ coating.

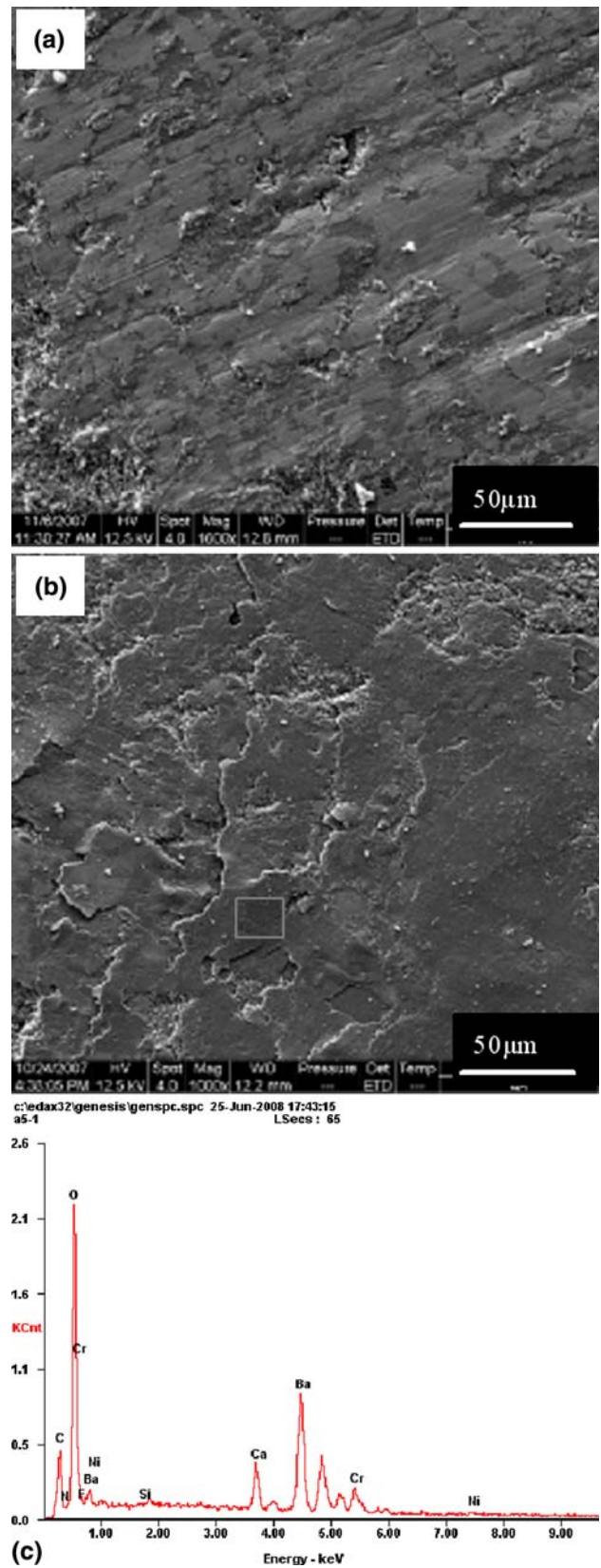
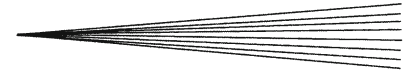


Fig. 7 Worn surfaces of the NiCr/Cr₃C₂-10%BaF₂CaF₂ coating (a) at room temperature, (b) at 500 °C, and (c) area EDS analysis of (b)



4. Conclusions

- (1) NiCr cladding can effectively reduce the difference of density and thermophysical properties between the feedstock components and protect the cores from oxidation and decarburization.
- (2) Adding BaF₂-CaF₂ eutectic in the NiCr/Cr₃C₂ feedstock has a negative effect on the mechanical properties of the coating, when the content exceeds 10%, the hardness and bond strength of the coating drop sharply.
- (3) BaF₂-CaF₂ eutectic can improve the tribological properties of the NiCr/Cr₃C₂ coating at high temperature. When the temperature reaches higher than 500 °C, the friction coefficient, the wear rates of NiCr/Cr₃C₂-10% BaF₂-CaF₂ coating, and its coupled Si₃N₄ ball are 20%, 40%, and 75% lower than those of the NiCr/Cr₃C₂ coating, respectively.
- (4) The dominated wear mechanism of the NiCr/Cr₃C₂-BaF₂-CaF₂ coating at room temperature is splats spallation and abrasive wear, whereas, when the temperature reaches 500 °C, it changes to plastic deformation and plowing by the Si₃N₄ ball.

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